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glycoside of Geraea canescens was purified by PC and obtained as a crystalline solid, R_f 60 (BAW), 64 (PhOH) and 44 (15% HOAc), λ_{max} 263, 272, 366; +NaOEt, 420; +NaOAc, 283; +H₃BO₃, 386; and +AlCl₃ 366, 414 nm. On acid or β -glucosidase hydrolysis, it gave galactose (with some glucose) and gossypetin 8-methyl ether (see Table 1). The aglycone was further confirmed by MS: M 332 (C₁₆H₁₂O₈ requires 332) (48%), M-15 (100%), M-29 (3%), M-43 (13%) and B ring fragment at 137 (12%).

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REFERENCES

- 1. Harborne, J. B. (1972) Recent Adv. Phytochem. 4, 107.
- 2. Gottlieb, O. R. (1975) in The Flavonoids (Harborne, J. B., Mabry, T. J. and Mabry, H. eds) 296-375. Chapman & Hall,

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- 3. Nielsen, J. G. (1970) Tetrahedron Letters 803.
- 4. Harborne, J. B. (1969) Phytochemistry 8, 177.
- Urschler, J. (1968) Phyton (Horn, Austria) 13, 15.
- 6. Bradner, N. R. and Brink, V. C. (1968) Can. J. Plant Sci.
- Wagner, H., Rüger, R., Maurer, G. and Farkas, L. (1977) Chem. Ber. 110, 737.
- Chumbalov, T. K. (1976) Khim. Prir. Soedin. 658.
- 9. Sporne, K. R. (1969) New Phytologist 68, 555.
- 10. Harborne, J. B. (1968) Phytochemistry 7, 1215.
- 11. Koehler, D. and Smith, D. M. (1977) Madrono in press.

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RARE METHYLATED FLAVONOLS FROM ANGELONIA GRANDIFLORA

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Key Word Index-Angelonia grandiflora; Scrophulariaceae; methylated flavonols; 3,4'-,3,7- and 7,4'-dimethylquercetin; 3,4'- and 7,4'-dimethylkaempferol.

We previously identified scutellarein 7,4'-dimethyl ether and 5-hydroxy-6,7,4'-trimethoxyflavone(salvigenin) from leaf tissue of Angelonia grandiflora [1]. In continuation of our investigation, we now report the isolation and identification of five rare methylated flavonols: 3,4'-, 3,7and 7,4'-dimethylquercetin and 3,4' and 7,4'-dimethylkaempferol. All five compounds are known from other sources: quercetin 3,4'-dimethyl ether from Baccharis sarothroides [2], 3,7-dimethylquercetin from Aeonium manriquiorum [3] and Larrea cuneifolia [4], 7,4'-dimethylquercetin from Phytolacca dioica [5], 3,4'-dimethylkaempferol from propolis [6] and Betula ermanii [7] and 7,4'-dimethylkaempferol from Cheilanthes farinosa [8]. Their present isolation together with our previous finding [1] of 7,4'-dimethylscutellarein and salvigenin reveals an unusual co-occurrence of rare methylated flavonols and flavones in A. grandiflora.

EXPERIMENTAL

Plant material. A voucher specimen of Angelonia grandiflora C. Mor. has been deposited at JIPMER.

Flavonoid identification. A hot C₆H₆ extract of dried leaf material was subjected to column chromatography over silicic acid using petrol (60-80°), petrol-C₆H₆, C₆H₆-CHCl₃ and CHCl3-MeOH as eluents. Petrol and petrol-C6H6 yielded carotenoids and triterpenes; other fractions gave flavone and flavonol derivatives including the five rare methylated flavonols, A-E.

Flavonol A (1). Major (CHCl₃- C_6H_6 , 3.1), mp 238-39°, UV—purple UV/NH₃—dull yellow; λ_{max} (nm) . 220 sh, 253, 268 sh, 353 (MeOH); 273, 321, 367 (NaOAc); 210 sh, 235 sh, 265, 301, 363, 395 sh (AlCl₃); 210 sh, 237 sh, 264, 301, 360, 396 sh (AlCl₃/HCl) and 235 sh 273, 325 sh, 410 (NaOMe); PMR signals $(CD_3SOCD_3, \delta \text{ values, ppm})$ at 12.60 (broad s, 1H, disappeared on D₂O exchange, 5-OH) 7.73 (d, 2 Hz, 1H, 2'-H) 7.67 (dd, 2 Hz and 9 Hz, 1H, 6'-H) 7.05(d, 9 Hz, 1H, 5'-H) 6.52 (unresolved d, 1H, 8-H) 6.28 (unresolved $d, 1\text{H}, 6\text{-H}) 3.96(s, 3\text{H}, 3\text{-OCH}_3)$ and 3.88 (s, 3H, 4'-OMe). Its MS exhibited the following ions: (m/e)330 (M⁺, 100%), 329 (M⁺-H, 40), 315 (M⁺-Me, 50), 299 (M⁺-OMe, 18), 297 (315-H₂O, 18), 287 (M⁺-MeCO, 90), 272 (287-Me, 18), 244 (M⁺-2MeCO, 35), 153 (RDA fragment $A_1 + H$, 50)*, 151 (fragment C,24) and 147 (RDA fragment B_1 – OMe, 30). It had R_f (× 100, Whatman No. 1, ascending, 30°) 16 (15% HOAc), 47 (30 % HOAc), 67 (50 % HOAc), 94 (BAW), 93 (phenol), 84 (Forestal) and 90 (tBAW). On methylation, it gave quercetin pentamethyl ether and on demethylation quercetin. It was

- $\begin{array}{ll} \textbf{1} & \textbf{R} = \textbf{OH}, \, \textbf{R}_3 = \textbf{H}, \, \textbf{R}_1 = \textbf{R}_2 = \textbf{Me} \\ \textbf{2} & \textbf{R} = \textbf{OH}, \, \textbf{R}_2 = \textbf{H}, \, \textbf{R}_1 = \textbf{R}_3 = \textbf{Me} \\ \end{array}$
- 3 R = OH, R₁ = H, R₂ = R₃ = Me 4 R = R₃ = H, R₁ = R₂ = Me 5 R = R₁ = H, R₂ = R₃ = Me

^{*}For explanations of A₁, B₁, etc, see [9].

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identified as 3,4'-dimethylquercetin and the identity confirmed by mmp and co-TLC with an authentic sample from *Baccharis sarothroides* [2].

Flavonol B (2). Major (CHCl₃-MeOH, 9:1) mp 250-51°, UV—purple; UV/NH₃—yellow; λ_{max} . 253, 268 sh, 290 sh, 354 (EtOH); 255, 269 sh, 325 sh, 360 (NaOAc); 256, 355, 405 sh (AlCl₃) and 270, 329, 411 (NaOEt). Its MS exhibited peaks at 330 (M^+ , 100%), 329 (M^+ -H, 50), 315 (M^+ -Me, 45), 301 (M^+ -CHO, 15), 299 (M+-OMe, 15), 287 (M+-COMe, 80), 244 (M+-2COMe, 22), 151 (RDA fragment A₁-Me, 25), 135 (A₁-OMe, 22) and 123 (151-CO, 18). PMR spectrum (90 MHz, CDCl₃) of its triacetyl derivative (mp 197–98°, λ_{max} :242, 323 (EtOH) had signals at 7.75 (d, 2 Hz, 1H, 2'-H) 7.68 (dd, 2 Hz and 9 Hz, 1H, 6'-H) 7.32 (d, 2 Hz, 1H, 8-H) 7.17 (d, 9 Hz, 1H, 5'-H) 6.84 (d, 2 Hz, 1H, 6-H) 3.96 (s, 3H, 7-OCH₃) 3.88 (s, 3H, 3-OCH₃) 2.48 (s, 3H, 5-OCOCH₃) and 2.36 (s, 6H, 3' and 4'-OCOCH₃). The MS of the acetate showed the parent ion at m/e 456 with fragment ions at 414 (M⁺-CH₂CO), 372 (M⁺-2CH₂CO, 100 %), 357 (372-Me), 330 (M+-3CH₂CO), 329 (357-CO), 315 (330-Me) and 287 (315-CO) confirming the dimethoxytriacetatoxyflavone structure. On PC, this flavonol had the same R, as A, but could be differentiated by TLC. It was identified as 3,7-dimethylquercetin.

Flavonol C (3). (CHCl₃-C₆H₆, 3:2), mp 198-200°, UV and UV/NH₃—yellow. λ_{max} . 254, 289, 363 (MeOH); 254 sh, 289, 368, 410 sh (NaOAc); 223, 265 sh, 311, 375 sh, 420 (AlCl₃) and 247, 289, 359, 447 (NaOMe). MS peaks at 330 (M⁺, 100%) 329, 316 (M⁺-14,70). R_f : 4, 20, 51, 91, 91, 79 and 86 in the above solvents. Demethylation gave quercetin and the compound was identified as 7.4′-dimethylquercetin by direct comparison with the compound synthesised [9] from penta-acetylquercetin and 2 equivalents of Me₂SO₄ and dry K₂CO₃ in Me₂CO for 10 hr.

Flavonol D (4). (CHCl₃–C₆H₆, 1.3), mp 188–90°, UV—purple; UV/NH₃—yellow; λ_{max} . 267, 295, 342 (MeOH); 274, 300, 358 (NaOAc); 232, 271 sh, 303, 348 (AlCl₃) and 248 sh, 262, 298, 370 (NaOMe). MS had a peak at 314 (M⁺, 100 %). R_{j} . 17, 50, 72, 94, 96, 92 and 95 in the above solvents. Its acetate had the M⁺ at 398 and on demethylation gave kaempferol. It was identified as 3,4′-dimethylkaempferol.

Flavonol E (5). (CHCl₃-C₆H₆, 1:3), mp 180-81°, UV and UV/NH₃—yellow; $\lambda_{\rm max}$. 225 sh, 270, 365 (MeOH); 256, 268, 383 (NaOAc); 268, 350, 420 (AlCl₃) and 266, 410 (NaOMe). MS peak at 314 (M⁺, 100%), 300 (M⁺-14,24). Its acetate had M⁺ at 398 and a prominent ion at 384 besides other expected ions; it gave kaempferol on demethylation and had R_i , 7, 18, 51, 85, 91, 79 and 87 in the above solvents; It was identified as 7,4'-dimethylkaempferol.

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REFERENCES

- Nair, A. G. R., Ramesh, P., Subramanian, S. S. and Joshi, B. S. (1976) Indian J. Chem. 14B, 463.
- 2. Kupchan, S. M. and Bauerschmidt, E. (1971) Phytochemistry 10, 664.
- Castillo, J. B., Gonzalez, A. G. and Eglinton. G (1968) Ann Quim. Spain. 64 193.
- Valesi, A. G., Rodriguez, E., Velde, V. G. and Mabry, T. J. (1972) Phytochemistry 11, 2821.
- Jurd, L. (1962) J. Org. Chem. 27, 1294.
 Propravko, S. A., Gurevich, A. I. and Kolosov, M. N. (1969) Khim. Prir. Soedin. 5, 476.
- 7. Woollenweber, E. (1971) Tetrahedron Letters 1767.
- 8. Erdtman, H., Novotny, L. and Romanuk, M. (1966) Tetrahedron 22, Suppl. 8, 71.
- Wagner, H. and Farkas, L. (1975) in The Flavonoids (Harborne, J. B., Mabry, T. J. and Mabry, H. eds) p. 155. Chapman & Hall, London.